

## Shannon entropy of 1-normalized electron density

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Received 16 November 1999; accepted 27 December 1999

The calculations of the 1-normalized Shannon entropy of Hartree-Fock electron density,  $S_\eta$ , and its variation with the number of electrons,  $[\partial S_\eta / \partial N]_Z$ , are reported for He-Xe atoms in the ground states. The N variation is shown as related to the Fukui function of the density functional theory and it becomes identically equal to zero if the electron density in N-electron system is described by the homogeneous electron gas.

During the recent years, the Shannon information entropy of electronic density,  $\rho(\mathbf{r})$ , defined as

$$S_p = - \int \rho(\mathbf{r}) \ln \rho(\mathbf{r}) \, d\mathbf{r} \quad \dots(1)$$

has been used in the analysis of several chemical and physical phenomena<sup>1-10</sup>. Substituting in Eq.(1),  $\rho(\mathbf{r}) / N$ , instead of  $\rho(\mathbf{r})$  defines the 1-normalised entropy,  $S_\eta$ , which is related to  $S_p$  according to

$$S_\eta = (S_p/N) + \ln N \quad \dots(2)$$

The quantity  $S_\eta$  is usually positive. In this communication we report on an interesting property of the derivative  $(\partial S_\eta / \partial N)_Z$  for N -electron systems. On differentiating  $S_\eta$  with respect to N, we obtain

$$[\partial S_\eta / \partial N]_Z = -(1/N^2) S_p - (1/N) \int (\partial \rho / \partial N)_Z \ln \rho(\mathbf{r}) \, d\mathbf{r} \quad \dots(3)$$

For a homogeneous electron gas,  $[\partial \rho / \partial N]_Z = \rho(\mathbf{r})/N$  which reduces Eq. (3) to  $[\partial S_\eta / \partial N]_Z = 0$ .

The quantity  $[\delta \rho(\mathbf{r}) / \delta N]_Z$  in equation (3) defines the Fukui function<sup>11</sup> - the chemical reactivity parameter in the density functional theory<sup>12</sup>. Thus, Eq. (3) provides a connection between an important reactivity parameter and the Shannon entropy of electron density.

We have used a modified Hartree-Fock 86 code<sup>13</sup> and obtained the numerical HF wave functions for He-Xe atoms in the ground state. These wave functions have been used to calculate  $S_p$ ,  $S_\eta$  and  $(\partial S_\eta / \partial N)_Z$  as given by

Table 1 - The results of 1-normalised Shannon entropy,  $S_\eta$  and its derivative with respect to N for He - Xe using HF densities. All quantities in a.u.

Z	Atom	State	$S_\eta$	$\partial S_\eta / \partial N$
2.0	He	<sup>1</sup> S	2.698	0.713
3.0	Li	<sup>2</sup> S	3.701	1.703
4.0	Be	<sup>1</sup> S	3.624	0.990
5.0	B	<sup>2</sup> P	3.405	0.776
6.0	C	<sup>3</sup> P	3.106	0.515
7.0	N	<sup>4</sup> S	2.802	0.485
8.0	O	<sup>3</sup> P	2.550	0.415
9.0	F	<sup>2</sup> P	2.299	0.298
10.0	Ne	<sup>1</sup> S	2.055	0.321
11.0	Na	<sup>2</sup> S	2.330	0.722
12.0	Mg	<sup>1</sup> S	2.395	0.565
13.0	Al	<sup>2</sup> P	2.446	0.529
14.0	Si	<sup>3</sup> P	2.419	0.403
15.0	P	<sup>4</sup> S	2.359	0.573
16.0	S	<sup>3</sup> P	2.299	0.328
17.0	Cl	<sup>2</sup> P	2.222	0.260
18.0	Ar	<sup>1</sup> S	2.134	0.261
19.0	K	<sup>2</sup> S	2.302	0.476
20.0	Ca	<sup>1</sup> S	2.363	0.399
21.0	Sc	<sup>2</sup> D	2.298	0.290
22.0	Ti	<sup>3</sup> F	2.219	0.257
23.0	V	<sup>4</sup> F	2.135	0.258
24.0	Cr	<sup>7</sup> S	1.956	0.269
25.0	Mn	<sup>6</sup> S	1.963	0.199
26.0	Fe	<sup>5</sup> D	1.882	0.210
27.0	Co	<sup>4</sup> F	1.800	0.210
28.0	Ni	<sup>3</sup> F	1.718	0.210
29.0	Cu	<sup>2</sup> S	1.563	0.299
30.0	Zn	<sup>1</sup> S	1.556	0.267
31.0	Ga	<sup>2</sup> P	1.574	0.278

Contd....

Table 1 - The results of 1-normalised Shannon entropy,  $S_\eta$  and its derivative with respect to  $N$  for He - Xe using HF densities. All quantities in a.u. (Contd...)

Z	Atom	State	$S_\eta$	$\partial S_\eta / \partial N$
32.0	Ge	<sup>4</sup> P	1.568	0.230
33.0	As	<sup>4</sup> S	1.550	0.222
34.0	Se	<sup>4</sup> P	1.534	0.205
35.0	Br	<sup>2</sup> P	1.511	0.174
36.0	Kr	<sup>1</sup> S	1.481	0.176
37.0	Rb	<sup>2</sup> S	1.576	0.286
38.0	Sr	<sup>1</sup> S	1.621	0.252
39.0	Y	<sup>2</sup> D	1.614	0.206
40.0	Zr	<sup>3</sup> F	1.595	0.186
41.0	Nb	<sup>6</sup> D	1.525	0.191
42.0	Mo	<sup>7</sup> S	1.491	0.179
43.0	Tc	<sup>6</sup> S	1.508	0.168
44.0	Ru	<sup>5</sup> F	1.431	0.166
45.0	Rh	<sup>4</sup> F	1.396	0.160
46.0	Pd	<sup>1</sup> S	1.305	0.155
47.0	Ag	<sup>2</sup> S	1.324	0.202
48.0	Cd	<sup>1</sup> S	1.331	0.185
49.0	In	<sup>2</sup> P	1.352	0.192
50.0	Sn	<sup>3</sup> P	1.357	0.175
51.0	Sb	<sup>4</sup> S	1.355	0.161
52.0	Te	<sup>1</sup> P	1.353	0.151
53.0	I	<sup>2</sup> P	1.347	0.143
54.0	Xe	<sup>1</sup> S	1.336	0.135

Eqs(1)-(3), respectively. In each case, we calculated  $S_\eta$  corresponding to  $N \pm 0.1$  and estimated the derivative  $(\partial S_\eta / \partial N)_z$  as numerical difference. The electron occupation was changed in the outermost orbital for each atom. Our results are presented in Table 1. Interestingly, the isonuclear variation with  $N$  of  $S_\eta$  attains a minimum at the halogen atoms. Within the local plasma model<sup>14</sup> Ho *et al.*<sup>15</sup> have shown that  $S_\eta$  can be used directly to estimate the mean excitation energy in the cases of H-Ar atoms. The present calculations can be used to carry out similar calculations in the cases of K-Xe atoms.

## Acknowledgement

KDS thanks the Third World Academy of Sciences for a travel grant under its South-South Fellowship Program. Financial support received from FONDECYT (Fondo de Ciencias y Tecnologia) Grant 1981 231 and DAAD, Bonn is gratefully acknowledged.

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